

## DIESEL FUEL REFORMING STRATEGY

## BACKGROUND

Alternative transportation fuels have been represented as enablers to reduce toxic emissions in comparison to those generated by conventional fuels. At the same time, tighter emission standards and significant innovation in catalyst formulations and engine controls has led to dramatic improvements in the low emission performance and robustness of gasoline and diesel engine systems. This has certainly reduced the environmental differential between optimized conventional and alternative fuel vehicle systems. However, many technical challenges remain to make the conventionally fueled internal combustion engine a nearly zero emission system having the efficiency necessary to make the vehicle commercially viable.

Alternative fuels cover a wide spectrum of potential environmental benefits, ranging from incremental toxic and carbon dioxide (CO<sub>2</sub>) emission improvements (reformulated gasoline, alcohols, etc.) to significant toxic and CO<sub>2</sub> emission improvements (natural gas, etc.). Hydrogen is clearly the ultimate environmental fuel, with potential as a nearly emission free internal combustion engine fuel (including CO<sub>2</sub> if it comes from a non-fossil source).

The automotive industry has made very significant progress in reducing automotive emissions. This has resulted in some added cost and complexity of engine management systems, yet those costs are offset by other advantages of computer controls: increased power density, fuel efficiency, drivability, reliability and real-time diagnostics.

Future initiatives to require zero emission vehicles appear to be taking us into a new regulatory paradigm where asymptotically smaller environmental benefits come at a very large incremental cost. Yet, even an "ultra low emission" certified vehicle can emit high emissions in limited extreme ambient and operating conditions or with failed or degraded components.

One approach to addressing the issue of emissions is the employment of fuel cells. A fuel cell is an energy conversion device that generates electricity and heat by electrochemically combining a gaseous fuel,

such as hydrogen, carbon monoxide, or a hydrocarbon, and an oxidant, such as air or oxygen, across an ion-conducting electrolyte. The fuel cell converts chemical energy into electrical energy. A fuel cell generally consists of two electrodes positioned on opposite sides of an electrolyte. The oxidant passes  
5 over the oxygen electrode (cathode) while the fuel passes over the fuel electrode (anode), generating electricity, water, and heat. There are several types of fuel cells, including proton exchange membrane (PEM) fuel cells and solid oxide fuel cells (SOFC).

In a SOFC, a fuel flows to the anode where it is oxidized by  
10 oxygen ions from the electrolyte, producing electrons that are released to the external circuit, and producing mostly water and carbon dioxide that are removed in the fuel flow stream. At the cathode, the oxidant accepts electrons from the external circuit to form oxygen ions. The oxygen ions migrate across the electrolyte to the anode. The flow of electrons through the external circuit  
15 provides for consumable or storable electricity. However, each individual electrochemical cell generates a relatively small voltage. Higher voltages are attained by electrically connecting a plurality of electrochemical cells in series to form a stack.

Using conventional fuels (i.e., gasoline, diesel) within a fuel cell  
20 damages the fuel cell from the deposition of carbon (or soot). Therefore, typical fuel sources for fuel cells are reformates, i.e., a form of purified hydrogen produced from a hydrocarbon fuel processed in a reformer. Three types of reformer technologies are typically employed (steam reformers, dry reformers, and partial oxidation reformers) to convert hydrocarbon fuel (methane, propane,  
25 natural gas, gasoline, etc) to hydrogen using water, carbon dioxide, and oxygen, respectfully, with byproducts including carbon dioxide and carbon monoxide, accordingly. These reformers operate at high temperatures (e.g., about 800°C or greater). Under steam reforming, an alcohol, such as methanol or ethanol, or a hydrocarbon, such as methane, gasoline or propane, is reacted with steam over  
30 a catalyst. Steam reforming requires elevated temperatures and produces primarily hydrogen and carbon dioxide. Some trace quantities of unreacted reactants and trace quantities of byproducts such as carbon monoxide also result from steam reforming.

Reforming of a fuel generally involves the conversion of a hydrocarbon fuel into separate components, such as hydrogen, carbon monoxide, and carbon dioxide. Generally, lighter hydrocarbons such as methane, ethane, and gasoline more easily reformed. Heavier hydrocarbon fuels, such as diesel, are inherently more difficult to reform because of its tendency to partially decompose prior to full vaporization. Thus, the amount of hydrogen or carbon monoxide recoverable is less than lighter hydrocarbon fuels. Also, diesel fuel is high in sulfur content, which may poison a fuel cell or result in pollution due to the production of sulfur oxides.

#### SUMMARY

The drawbacks and disadvantages of the prior art are overcome by the diesel fuel reforming strategy.

A method of using a diesel reforming strategy is disclosed. The method comprises supplying diesel fuel to a fractional distillation device. The diesel fuel is fractionally distilled to produce a light fuel stream and a heavy fuel stream. The light fuel stream is reformed in a reformer to produce a reformat.

A diesel fuel reforming system is also disclosed. The diesel fuel reforming system comprises a fractional distillation device in fluid communication with a supply of diesel fuel. A reformer is in fluid communication with the fractional distillation device. A fuel cell stack is in fluid communication with the reformer.

A method of making an apparatus for a diesel reforming strategy is disclosed. The method comprises a reformer in fluid communication with a fuel cell stack, the reformer in fluid communication with a fractional distillation device, and the fractional distillation device in fluid communication with a supply of diesel fuel.

A method of using a fuel cell system is disclosed. The method comprises supplying diesel fuel to a fractional distillation device. The diesel fuel is fractionally distilled to produce a light fuel stream and a heavy fuel stream. The light fuel stream is reformed in a reformer to produce a reformat. The reformat is utilized in a fuel cell stack to produce electricity.

A fuel cell system for diesel fuel reforming is also disclosed. The system comprises a means for fractionally distilling a supply of diesel fuel

to produce a light fuel stream and a heavy fuel stream. A means for reforming the light fuel stream is utilized to produce a reformat, such that the means for reforming is disposed in fluid communication with the means for fractionally distilling. A means for producing electricity from the reformat is disposed in fluid communication with the means for reforming.

The above described and other features are exemplified by the following figures and detailed description.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Referring now to the figure, which is meant to be exemplary not limiting.

Figure 1 is a perspective view of an exemplary diesel fuel reforming strategy.

#### DETAILED DESCRIPTION

A fuel cell system may be utilized with an engine to operate a vehicle. The power created from the engine and the fuel cell can propel a vehicle, as well as providing electricity and heat for the auxiliary systems. To create this power, a fuel cell needs to be supplied with a fuel. Generally, lightweight fuels are supplied to a reformer to create a fuel usable in the fuel cell (commonly known as a "reformat"). To prepare a reformat from diesel fuel, the diesel fuel can be fractionally distilled to produce a stable light fuel vapor stream and a heavy fuel stream. The light fuel vapor stream is then processed in a reformer to produce a reformat suitable for use in a fuel cell. The diesel reforming strategy for a fuel cell system described herein is for any fuel cell system, including SOFC systems and PEM fuel cell systems.

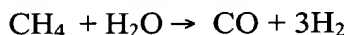
Referring to Figure 1, the apparatus for preparing a reformat from diesel fuel comprises a fractional distillation device 10 and a reformer 20. The fractional distillation device 10 distills a fuel 12, such as diesel, into a light fuel vapor stream 14 and a heavy fuel stream 16. A reformer 20 produces a reformat 22, such that the reformer 20 is adapted to receive the light fuel vapor stream 14. Possible fuels which can be processed include commercial diesel fuels, military diesel fuels, blended diesel fuels containing a larger than normal

During use, the diesel fuel would be introduced to the fractional distillation apparatus where the components of different boiling points are separated from a liquid mixture by heating the diesel fuel and vaporizing the liquid. The vapor is then passed through a fractionating column. Fractional distillation can be carried out by methods well known in the art.

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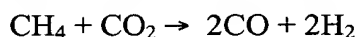
the downstream device or reformer. The light fuel vapor stream 14 can be processed in a reformer 20 that utilizes a steam reforming, a partial oxidation, or dry reforming process to produce a reformat 22, with a steam reforming process preferred. Steam reforming systems use a light fuel stream and steam (H<sub>2</sub>O) that is reacted in heated tubes filled with catalysts to convert the hydrocarbons into principally hydrogen and carbon monoxide. An example of the steam reforming reaction using methane as the fuel is as follows:



Partial oxidation reformers are based on substoichiometric combustion to achieve the temperatures necessary to reform the hydrocarbon fuel. Decomposition of the fuel to primarily hydrogen and carbon monoxide occurs through thermal reactions at high temperatures of about 700°C to about 1,000°C. The heat required to drive the reaction is typically supplied by burning a portion of the fuel. Catalysts have been used with partial oxidation systems (catalytic partial oxidation) to promote conversion of various low sulfur fuels into synthesis gas. The use of a catalyst can result in acceleration of the reforming reactions and can provide this effect at lower reaction temperatures than those that would otherwise be required in the absence of a catalyst. An example of the partial oxidation reforming reaction is as follows:



Dry reforming involves the creation of hydrogen and carbon monoxide in the absence of water, for example using carbon dioxide. An example of the dry reforming reaction is depicted in the following reaction:



Both steam reforming and dry reforming are endothermic processes, while partial oxidation is an exothermic process. A SOFC will utilize the hydrogen (i.e., reformat) and carbon monoxide (CO) as a fuel.

One advantage to this distillation process is the ability to use commercial or military diesel fuels. These fuels are generally difficult to reform because of the tendency for the fuels to partially decompose prior to full vaporization. The fact that these fuels often have a high sulfur concentration is a further problem with the use in a fuel cell since sulfur is a fuel cell poison. However, this process reforms the lighter portion of the diesel fuel, removes at

least some, and preferably all, of the sulfur, and optionally utilizes the heavy fuel to heat various system components.

Commercial diesel fuels tend to be hydro-treated and, consequentially, the light fuel vapor stream from the distilled hydro-treated diesel fuel can have even lower concentrations of sulfur, compared to non-hydrotreated diesel fuel. Consequently, it is ideal to use hydrotreated commercial fuels with a fuel cell since sulfur can poison a fuel cell. Yet, the employment of the fractional distillation process enables the use of a variety of fuels including non-hydrotreated diesel fuels, which may be more readily available in third world countries.

When fractionally distilling commercial diesel fuels, the distilled reformat available to the fuel cell is inherently lower in sulfur compared to an approach where all of the fuel passes through the anode. Additionally, other contaminants, such as organometallic compounds and polyaromatic hydrocarbons, are also present in commercial fuels and can foul a fuel cell. By distilling the diesel fuel into a light fuel stream and a heavy fuel stream, these contaminants are similarly removed via the heavy fuel stream. This method of reforming heavier hydrocarbon fuels allows for the full mixing of air and or steam reactants and is much less prone to sooting and deterioration of the fuel cell catalysts.

While the invention has been described with reference to an exemplary embodiment, it will be understood by those skilled in the art that various changes may be made and equivalents may be substituted for elements thereof without departing from the scope of the invention. In addition, many modifications may be made to adapt a particular situation or material to the teachings of the invention without departing from the essential scope thereof. Therefore, it is intended that the invention not be limited to the particular embodiment disclosed as the best mode contemplated for carrying out this invention, but that the invention will include all embodiments falling within the scope of the appended claims.

What is claimed is: